

# Photochemistry of Some Codeinones and Morphinones\*

H. R. Lotfy<sup>1</sup>, A. G. Schultz<sup>†2</sup>, and M. A. Metwally<sup>3</sup>

<sup>1</sup> El-Khiaria, El-Mansoura, Egypt

<sup>2</sup> Department of Chemistry, RPI (Rensselaer Polytechnic Institute), Troy, New York, 12180-3590, USA

<sup>3</sup> Department of Chemistry, Faculty of Science, University of Mansoura, Egypt

Received February 12, 2003

**Abstract**—Irradiation of 5-benzyl-, 5-methoxymethyl-, and 5-allyl-substituted codeinones and morphinones with UV light ( $\lambda$  366 nm) leads to formation of the corresponding 18 $\beta$ -phenyl, 18 $\beta$ -methoxy, and 18 $\beta$ -vinyl derivatives of 4,5 $\alpha$ -epoxymethanomorphin-7-en-6-one.

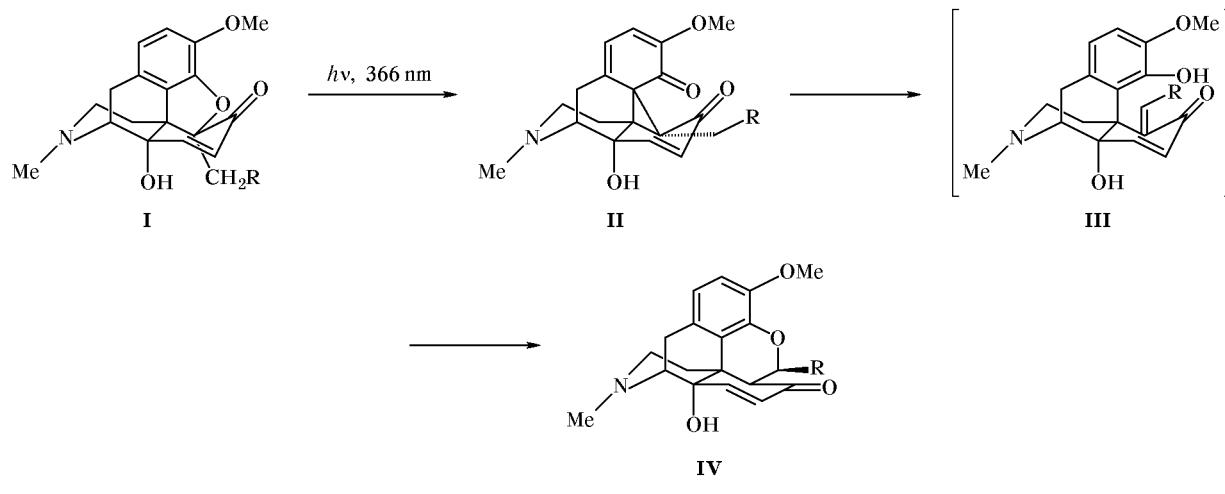
Schultz *et al.* previously reported [1] on the photochemistry of codeinone derivatives, which involved unique photoreactivity of the dihydrobenzofuran ring system [2–4]. The photorearrangements were initiated by UV irradiation at  $\lambda$  366 nm [5]. In the present work we tried to extend analogous transformations to some other 5-substituted derivatives.

Irradiation of a solution of 5-benzyl-14-hydroxycodeinone (**I**) [6] in methanol at  $\lambda$  366 nm for 75 min, followed by chromatography on a silica gel plate, gave a colorless product together with the unreacted initial compound. By recrystallization from hexane we

isolated benzopyran derivative **IV** as colorless crystals (yield 85%, calculated on the reacted **I**).

A plausible mechanism [1] of the photochemical rearrangement of 5-benzyl-14-hydroxycodeinone (**I**) into benzopyran (**IV**) includes isomerization of the initial compound into cyclopropane-spiro-2,4-cyclohexadienone **II** (Scheme 1). The latter undergoes unusual Claisen rearrangement (C $\rightarrow$ O migration of hydrogen) involving the 5-benzyl substituent to give intermediate dienone **III** which was not detected in the reaction mixture. Intramolecular Michael addition in **III** converts it to benzopyran **IV**. The formation

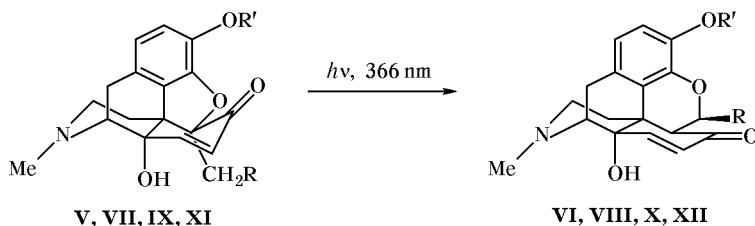
Scheme 1.



**I–IV**, R = Ph.

† Deceased.

\* The original article was submitted in English.

**Scheme 2.**

**V, VI, R = MeO; VII, VIII, XI, XII, R = CH<sub>2</sub>=CH; IX, X, R = Ph; V–VIII, R' = Me; IX–XII, R' = H.**

of product **IV** is the first demonstration that the photo-rearrangement reported for 5-methylcodeinone [1] can be extended to include other C<sup>5</sup>-substituted codeinone derivatives.

Irradiation of 5-methoxymethylcodeinone (**V**) in methanol at  $\lambda$  366 nm for 3 h (further irradiation resulted in decomposition) and subsequent chromatographic separation afforded 25% of benzopyran **VI** as a brown oil (Scheme 2). Compound **VIII** was obtained by irradiation of a methanolic solution of 5-allyl-14-hydroxocodeinone (**VII**) at  $\lambda$  366 nm over a period of 6 h. The yield of **VIII** was 60% (calculated on the reacted initial compound). Likewise, 5-benzyl-14-hydroxymorphinone (**IX**) and 5-allyl-14-hydroxymorphinone (**XI**) were converted into benzopyran derivatives **X** (yield 72%) and **XII** (51%), respectively, by irradiation at  $\lambda$  366 nm for 2 and 6 h.

## EXPERIMENTAL

The <sup>1</sup>H NMR spectra were recorded on Varian EX-200 (200 MHz) and Unity 500 (500 MHz) spectrometers using chloroform-d or benzene-d<sub>6</sub> as solvent and internal reference. The <sup>13</sup>C NMR spectra were obtained on a Varian Unity 500 instrument in chloroform. The IR spectra were measured on Perkin-Elmer Model 298 and FTIR spectrometers. The chemical ionization mass spectra were run on a Hewlett-Packard 5987A GC-MS system with the use of isobutene as chemical ionization gas.

**General procedure for photochemical reactions.** Appropriate substrate was dissolved in a spectrophotometric grade solvent to a required concentration (see below), and the solution was placed in a 5- or 30-ml test tube. The solution was irradiated with a Hanovia medium-pressure mercury arc lamp (450 W), which was placed in a uranyl glass sleeve ( $\lambda$  366 nm) cooled with water. The entire apparatus and the sample were immersed in a water-cooled bath, and the sample was irradiated for a time indicated below. The solvent was removed under reduced pressure to afford the crude product.

**14-Hydroxy-3-methoxy-17-methyl-18 $\beta$ -phenyl-4,5 $\alpha$ -epoxymethanomorphin-7-en-6-one (**IV**).** A solution of 0.012 g (0.030 mmol) of 5-benzyl-14-hydroxocodeinone in 5 ml of methanol was irradiated at  $\lambda$  366 nm over a period of 75 min. Removal of the solvent left a brown gum which was subjected to thin-layer chromatography on silica gel using chloroform-methanol (9:1) as eluent to isolate a colorless product and 0.006 g of the starting material. The product was recrystallized from hexane. Yield 0.0051 g (85% on the reacted initial compound). Colorless crystals with mp 180–182°C. <sup>1</sup>H NMR spectrum (CDCl<sub>3</sub>, 500 MHz),  $\delta$ , ppm: 1.91 d.d (1H, *J* = 3.6, 3.5 Hz), 2.44 m (2H), 2.48 s (3H, NMe), 2.61 d (1H, *J* = 6.6 Hz), 2.72 d.d (1H, *J* = 12, 11 Hz), 2.97 d (1H, *J* = 6.1 Hz), 3.29 d (1H, *J* = 18.6 Hz), 3.55 d (1H, *J* = 2.4 Hz), 3.84 s (3H, 3-OMe), 4.8 br.s (1H), 5.68 d (1H, *J* = 1.9 Hz), 5.71 d (1H, *J* = 9.8 Hz), 6.43 d (1H, *J* = 10 Hz), 6.61 d (1H, *J* = 8.3 Hz), 6.72 d (1H, *J* = 8.3 Hz), 7.26 m (1H), 7.40 t (2H, *J* = 8.1, 7.5 Hz), 7.59 d (2H, *J* = 7.5 Hz). <sup>13</sup>C NMR spectrum (CDCl<sub>3</sub>, 500 MHz),  $\delta$ <sub>C</sub>, ppm: 194.16, 146.87, 144.67, 140.88, 132.11, 127.88, 126.48, 124.81, 122.08, 118.14, 111.16, 72.35, 67.43, 61.67, 56.27, 48.19, 45.28, 42.41, 38.96, 34.15, 23.79. IR spectrum (CHCl<sub>3</sub>),  $\nu$ , cm<sup>-1</sup>: 3500, 1700. Mass spectrum, *m/z* (*I*<sub>rel</sub>, %): 404 (*M*<sup>+</sup> + 1, 100), 388 (10), 373 (2), 298 (12).

**14-Hydroxy-3,18 $\beta$ -dimethoxy-17-methyl-4,5 $\alpha$ -epoxymethanomorphin-7-en-6-one (**VI**).** A solution of 20 mg (0.056 mmol) of 14-hydroxy-5-methoxymethylcodeinone in methanol (6 ml) was irradiated at  $\lambda$  366 nm over a period of 3 h. The solvent was removed under reduced pressure, and the residue was subjected to chromatography on silica gel using chloroform-methanol (9:1) as eluent to isolate 5 mg (25%) of the product as a brown oil. <sup>1</sup>H NMR spectrum (CDCl<sub>3</sub>, 200 MHz),  $\delta$ , ppm: 1.55 br.s (1H), 2.2–2.7 m (4H), 2.4 s (3H, NMe), 2.91 d (1H, *J* = 4 Hz), 3.25 m (2H), 3.61 s (3H, OMe), 3.8 s (3H, OMe), 3.8 d (1H, *J* = 3.8 Hz), 5.97 d (1H, *J* = 10.09 Hz), 6.03 d (1H, *J* = 1.92 Hz), 6.6 m (3H). IR spectrum (CHCl<sub>3</sub>),  $\nu$ , cm<sup>-1</sup>: 1675, 3380. Mass spectrum, *m/z*

( $I_{\text{rel}}$ , %): 358 ( $M^+ + 1$ , 45), 326 (100), 324 (10), 310 (12).

**14-Hydroxy-3-methoxy-17-methyl-18 $\beta$ -vinyl-4,5 $\alpha$ -epoxymethanomorphin-7-en-6-one (VIII).** A solution of 20 mg (0.057 mmol) of 5-allyl-14-hydroxocodeinone in 6 ml of methanol was irradiated at  $\lambda$  366 nm over a period of 6 h. The solvent was removed under reduced pressure, and the residue was subjected to chromatography on silica gel using chloroform–methanol (10:1) as eluent to isolate 10 mg of the initial compound and 6 mg (60% on the reacted starting material) of product **VIII** as a yellow oil.  $^1\text{H}$  NMR spectrum ( $\text{CDCl}_3$ ),  $\delta$ , ppm: 1.73 d.d (1H,  $J = 3.6, 1.8$  Hz), 2.3 m (1H), 2.38 s (3H, NMe), 2.5–3.0 m (3H), 3.05 d (1H,  $J = 2$  Hz), 3.2 d (1H,  $J = 20$  Hz), 3.7 s (3H, OMe), 3.75 s (1H), 4.95 d (1H,  $J = 4$  Hz), 5.24 m (2H), 5.8 d (1H,  $J = 10$  Hz), 6.4 d (1H,  $J = 10$  Hz), 6.55 d.d (2H,  $J = 8.8$  Hz), 6.8 m (1H). IR spectrum ( $\text{CHCl}_3$ ),  $\nu$ ,  $\text{cm}^{-1}$ : 3385, 1680. Mass spectrum,  $m/z$  ( $I_{\text{rel}}$ , %): 354 ( $M^+ + 1$ , 100), 338 (7), 311 (8), 276 (15).

**3,14-Dihydroxy-17-methyl-18 $\beta$ -phenyl-4,5 $\alpha$ -epoxymethanomorphin-7-en-6-one (X).** A solution of 0.012 g (0.031 mmol) of 5-benzyl-14-hydroxymorphin-6-one (**IX**) in 6 ml of methanol was irradiated at  $\lambda$  366 nm over a period of 2 h. The solvent was removed to obtain a brown gum which was subjected to chromatography on a silica gel plate using methylene chloride–methanol (9:1) as eluent. We isolated 0.003 g of the unreacted starting material and 0.0065 g (72% on the reacted initial compound) of product **X** as a pink foam.  $^1\text{H}$  NMR spectrum ( $\text{CDCl}_3$ , 500 MHz),  $\delta$ , ppm: 1.9 m (1H), 2.5 m (1H), 2.49 s (3H, NMe), 2.3–2.6 m (3H), 2.97 d (1H,  $J = 5.6$  Hz), 3.28 d (1H,  $J = 18.6$  Hz), 3.52 d (1H,  $J = 2.2$  Hz), 5.68 s (1H), 5.73 d (1H,  $J = 9.8$  Hz), 6.45 d (1H,  $J = 10$  Hz), 6.6 m (2H), 7.27–7.52 m (5H). IR spectrum

( $\text{CHCl}_3$ ),  $\nu$ ,  $\text{cm}^{-1}$ : 3350 br, 1680. Mass spectrum,  $m/z$  ( $I_{\text{rel}}$ , %): 390 ( $M^+ + 1$ , 40), 313 (16), 270 (21), 254 (100).

**3,14-Dihydroxy-17-methyl-18 $\beta$ -vinyl-4,5 $\alpha$ -epoxymethanomorphin-7-en-6-one (XII).** A solution of 10 mg (0.029 mmol) of 5-allyl-14-hydroxymorphin-6-one (**XI**) in 6 ml of methanol was irradiated at  $\lambda$  366 nm over a period of 6 h. The solvent was removed to afford a brown residue which was subjected to chromatography on a silica gel plate using methylene chloride–methanol (9:1) to isolate 0.004 g of the unreacted initial compound and 0.0031 g (51% on the reacted **XI**) of compound **XII** as a pale yellow oil.  $^1\text{H}$  NMR spectrum ( $\text{CDCl}_3$ ),  $\delta$ , ppm: 1.73 m (1H), 2.3–2.9 m (4H), 2.38 s (3H, NMe), 3.05 d (1H,  $J = 2$  Hz), 3.2 d (1H),  $J = 20$  Hz, 3.75 s (1H), 5.0 d (1H,  $J = 4$  Hz), 5.24 m (2H), 5.8 d (1H,  $J = 10$  Hz), 6.4 d (1H,  $J = 10$  Hz), 6.5–6.9 m (3H). IR spectrum ( $\text{CDCl}_3$ ),  $\nu$ ,  $\text{cm}^{-1}$ : 3350 br, 1680.

The authors thank the World Laboratory of Switzerland for some financial support.

## REFERENCES

- Schultz, A.G., Green, N.J., Archer, S., and Tham, F.S., *J. Am. Chem. Soc.*, 1991, vol. 113, p. 6280.
- Schultz, A.G., Napier, J.J., and Lee, R., *J. Org. Chem.*, 1979, vol. 44, p. 663.
- Schultz, A.G., Ranganthan, R., and Kulkarni, Y.S., *Tetrahedron Lett.*, 1982, vol. 23, p. 4527.
- Schultz, A.G., Napier, J.J., and Sundaraman, P., *J. Am. Chem. Soc.*, 1984, vol. 106, p. 3590.
- Glasel, J.A. and Venn, R.F., *Life Sci.*, 1981, vol. 29, p. 221.
- Lotfy, H.R., Schultz, A.G., and Metwally, M.A., *Russ. J. Org. Chem.*, 2003, vol. 39, p. 1256.